

Nickel oxide performance as anode material for Lithium Ion Batteries

M. Ortiz^{1,2}, A. Visintin¹ and S. Real¹

¹*Instituto de Investigaciones Fisicoquímicas Teóricas y Aplicadas (INIFTA), Facultad de Ciencias Exactas, UNLP, CCT La Plata-CONICET, Suc. 4, C.C. 16, 1900 La Plata Argentina.*

²*Centro de Investigación y Desarrollo en Ciencia y Tecnología de los Materiales (CITEMA), Facultad Regional La Plata, Universidad Tecnológica Nacional (FRLP- UTN), Calle 60 y 124, La Plata, Argentina*

Lithium ion batteries, as a rechargeable power source, have attracted much more attention due to their extensive applications in portable electronic devices and electric vehicles. Although most commercial Li-ion batteries use graphite as anode, a variety of materials have been investigated in order to increase the cell capacity, and therefore its specific energy. Some of these materials are transition metal oxides, which are able to store more Li per gram than graphite and to improve their specific capacities. In this respect, special attention has been given to Fe, Co, and Ni based oxides [1-3].

Particularly, NiO is one of the promising anode materials for Li-ion batteries because of its low cost, environmental friendless and high theoretical capacity values (718 mA h g⁻¹ for 2Li⁺ per NiO). Various NiO components with different structures such as mesoporous, nanosheet, networks, nanowall, nanotube and hollow microsphere have been successfully fabricated [4-5].

In this work, the preparation and characterization of nickel oxide as anodes materials in lithium-ion batteries are presented. Two processes are involved in the synthetic procedure; in the first step the nickel hydroxide was obtained by hydrothermal synthesis (4h, 180°C) and then the precipitated was washed with distilled water to remove the residual species. The second step consists of the material calcinations in air at 300°C, for 4 (NiO-4h) and 24(NiO-24h) hours.

The structural characteristics and electrochemical properties of the obtained nickel oxides are subsequently investigated by optical and electrochemical techniques such as: FTIR, SEM, charge-discharge cycles, galvanostatic discharge at different currents and cyclic voltammetry.

The anode materials (NiO-4h and NiO-24h) were synthesized via a facile two-step route and exhibit a satisfactory specific capacity, cyclability and rate capability (Figure 1). These results indicate that the studied electrodes could be suitable as anodes in lithium ion batteries applications.

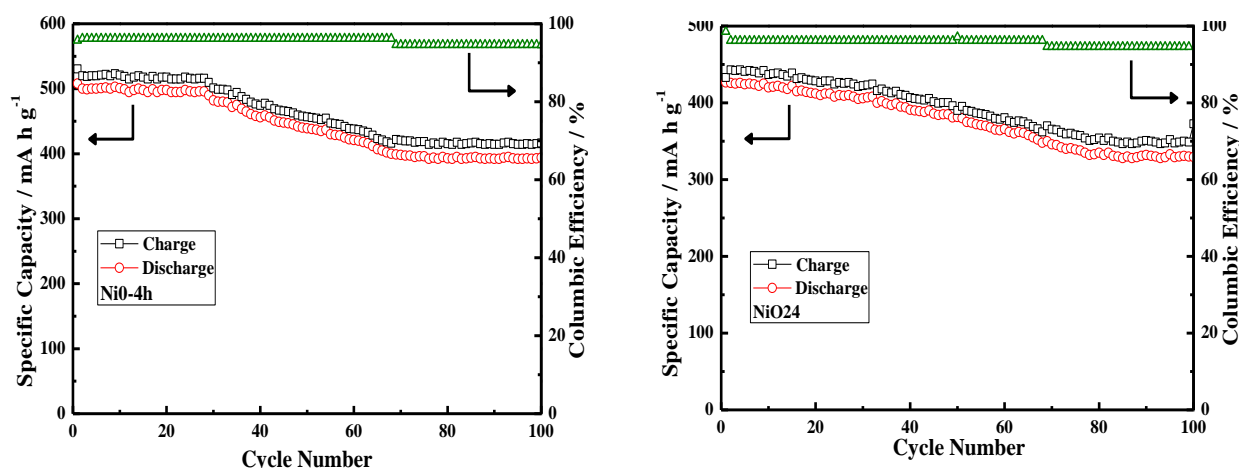


Figure 1. Cycle capacity of the NiO-4h and NiO-24h materials, at 0.5C

Keywords: Nickel oxide, anode materials, lithium ion batteries.

Acknowledgments: *This work was financially supported by Agencia Nacional de Promoción Científica y Tecnológica (ANPCyT), Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET), and Universidad Tecnológica Nacional (UTN).*

References:

- [1] P. Poizot, S. Laruelle, S. Grugeon, L. Dupont, J.-M. Tarascon, Nano-sized transition-metal oxides as negative-electrode materials for lithium-ion batteries, *Nature* 407 (2000) 496.
- [2] Yan Liu, Xiaogang Zhang, Effect of calcination temperature on the morphology and electrochemical properties of Co_3O_4 for lithium-ion battery, *Electrochim. Acta* 54 (2009) 4180.
- [3] X.H. Huang, J.P. Tu, X.H. Xia, X.L. Wang, J.Y. Xiang, L. Zhang, Y. Zhou, Morphology effect on the electrochemical performance of NiO films as anodes for lithium ion batteries, *J Power Sources* 188 (2009) 588.
- [5] S.A. Needham, G.X. Wang, H.K. Liu, Synthesis of NiO nanotubes for use as negative electrodes in lithium ion batteries, *J. Power Sources* 159 (2006) 254.
- [6] X.H. Huang, J.P. Tu, C.Q. Zhang, F. Zhou, Hollow microspheres of NiO as anode materials for lithium-ion batteries, *Electrochim. Acta* 55 (2010) 898.