Membrane-based processes: optimization of hydrogen separation by minimization of power, membrane area, and cost

Patricia L. Mores ^{1,+}, Ana M. Arias ^{1,+}, Nicolás J. Scenna ¹, José A. Caballero ², Sergio F. Mussati ³, and Miguel C. Mussati ^{3,*}

6 ¹ CAIMI Centro de Aplicaciones Informáticas y Modelado en Ingeniería, Universidad Tecnológica Nacional,
 7 Facultad Regional Rosario, Zeballos 1346, S2000BQA Rosario, Argentina

8 ² Department of Chemical Engineering, University of Alicante, Apartado de correos 99, 03080 Alicante, Spain

9 ³ INGAR Instituto de Desarrollo y Diseño (CONICET-UTN), Avellaneda 3657, S3002GJC Santa Fe, Argentina

- 10 ⁺ The authors have contributed equally.
- 11 * Correspondence: mmussati@santafe-conicet.gov.ar; Tel.: +54-342-453-4451
- 12 Received: date; Accepted: date; Published: date

13 Abstract: This paper deals with the optimization of two-stage membrane systems for H₂ separation 14 from a CO₂/CO/H₂/N₂ gas mixture in order to attain high values of both H₂ recovery and H₂ 15 product purity simultaneously at the minimum total annual cost. For a given H₂ recovery level of 16 90%, cost optimizations are performed and discussed for desired H₂ product purity values ranging 17 between 0.90 and 0.95 mole fraction. The results showed that the minimal total annual cost 18 increases exponentially with increasing H₂ product purity levels, and that the contribution of the 19 operating expenditures is more significant than the contribution of the annualized capital 20 expenditures (approximately 62% and 38%, respectively). This percentage contribution remains 21 almost constant in the studied range of H_2 product purity values. In addition, the optimal designs 22 (process-units sizes and operating conditions) obtained for different H₂ product purity levels and 23 90% H₂ recovery are discussed in detail. It was found an increment of H₂ product purity in 0.01 24 determines different percentage variations in costs depending on the purity level itself. For 25 instance, an increment of H₂ purity from 0.90 to 0.91 implies an increase of the total annual cost in 26 0.03739 M\$ yr.¹ (2.1%) while the same increment from 0.94 to 0.95 implies an increase in 0.17274 27 M\$ yr.⁻¹ (8.4%). Similarly, the optimal trade-offs existing between process variables –like the total 28 membrane area and the total electric power- depend on the specified H₂ product purity level. 29 Finally, the influence of two other single objective functions on the optimal designs of the studied 30 membrane process is analyzed: minimization of the total membrane area and minimization of the 31 total power. For the same design specifications (H_2 recovery of 90% and H_2 product purity of 0.90), 32 a minimal total annual cost of 1.76421 M\$ is obtained by cost minimization, which is about 4.7% 33 and 16.7% lower than the value obtained by minimization of the total membrane area and the total 34 electric power, respectively. From a comparison of the optimization results obtained for the three 35 objective functions, a strategy to systematically and rationally provide 'good' lower and upper 36 bounds for model variables and initial guess values to solve the cost minimization problem by 37 means of global optimization algorithms is proposed, which can be straightforward applied to 38 other processes. From the process system engineering perspective, the proposed optimization 39 model constitutes a valuable decision-support tool to design, simulate, and optimize two-stage 40 membrane processes for hydrogen separation, as well as to elucidate the exiting techno-economic 41 trade-offs that are difficult to distinguish at first glance.

42 Keywords: H₂ separation; membranes; multi-stage process; optimization; design; operation; cost;
 43 membrane area; energy; mathematical programming; NLP; GAMS.

44

Together with material science, the mathematical modeling and algorithmic optimization of both the membrane module and the whole membrane-based process are of main concern for improving the performance of this separation technology. Certainly, they can be easily used for testing and providing valuable information about the sizes of the process units as well as the operating conditions of the entire process flow sheet in a short time [1,2].

51 Depending on the degrees of freedom of the system -the difference between the number of 52 model variables and the number of equality constraints- a mathematical model of the process can be 53 used for two purposes: a) process simulation, when the degrees of freedom is null, and b) process 54 optimization, when the degrees of freedom is higher than 1. In literature, several authors have 55 addressed the optimization of membrane-based processes in several applications by employing 56 simulation-based optimization procedures [1,3-6] and other ones by using rigorous optimization 57 algorithms [7–10]. Without pretending to be an exhaustive review of the state-of-the-art of the types 58 of models and solution strategies -and independently of the membrane materials, geometry and 59 flow pattern, and the processed gas mixtures- some articles are briefly mentioned next that have 60 contributed to membrane separation from the process engineering or process system engineering 61 perspectives. For instance, Xu et al. [3] studied the potential applications of membrane-based 62 processes for hydrogen purification and pre-combustion CO₂ capture. They investigated single-stage 63 and two-stage configurations, and two membrane types: CO₂ selective membranes and H₂ selective 64 membranes (HSMs). Among other results, the authors found that a minimum cost selectivity can be 65 obtained by fixing the membrane permeability along with the H₂ product purity level. 66 Another important result indicated that it is difficult to reach a stable operation mode of the 67 two-stage system with HSM because it is strongly influenced by the variation of the operating 68 conditions. The authors highlighted the need of further investigation in this matter.

Giordano et al. [4] studied a single-stage unit to capture the CO₂ generated in a coal-fired power plant. They investigated the influence of the membrane operating temperature on the CO₂ capture considering two membrane materials with different gas separation properties (permeability and selectivity). Also, the authors considered feed compression and permeate vacuum pumping. Results showed that the increase in the operating temperature determines a decrease in the permeate CO₂ purity and an increase in the electric power requirement.

Ahmad et al. [1] implemented in Aspen HYSYS a two-dimensional mathematical model of a membrane-based process to study the sweetening of natural gas by capturing CO₂. They investigated several design configurations, from single to multiple stages including recycle streams and considering cross-flow pattern. The minimum gas processing cost was obtained with a two-stage membrane configuration where the retentate obtained in the first stage is fed to the second stage and the permeate obtained in the second stage is sent back to the first one.

By using a nonlinear mathematical programming (NLP) model –implemented in GAMS software–, Zarca et al. [7] evaluated a two-stage membrane process for H₂ recovery from the tail gas generated in carbon black manufacturing process, considering two types of membranes: polymeric membranes and ionic liquid-based membranes. Results show that ionic liquid-based membranes are promising not only to achieve a H₂-rich syngas stream at a minimal cost but also to mitigate CO₂ emissions.

87 Ohs et al. [10] proposed a mixed-integer nonlinear mathematical programming (MINLP) model 88 to address the optimal membrane cascade for N_2 removal from natural gas. As a result, the optimal 89 process configuration with the corresponding membrane areas, pressures of the feed and permeate 90 streams, including the selection of candidate recycle streams, were obtained. They considered 91 CH4-selective membranes only, N2-selective membranes only, and combinations of both. They found 92 that about 40% of the cost can be saved if a combination of CH4 and N2-selective membranes is used. 93 This work focuses on the rigorous optimization of membrane-based processes to separate H₂ 94 from a CO₂/CO/H₂/N₂ gas mixture generated in hydrocarbons processing plants to attain desired H₂ 95 recovery and product purity levels by minimization of cost, membrane area, and energy, separately. 96 To this end, three NLP problems are solved using the algebraic equation-oriented optimization tool

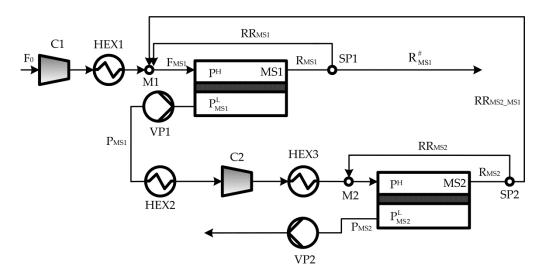
97 GAMS.

The paper is organized as follows. Section 2 describes the studied two-stage membrane process configuration for hydrogen separation. Section 3 states the optimization problem to be solved. Section 4 summarizes the main model assumptions and considerations and presents the mathematical model employed in this research. Section 5 discusses the optimization results obtained for different H₂ purity target levels and compares the optimal solutions obtained considering different objective functions in the optimization problem. Finally, Section 6 draws the conclusions of this work.

105 2. Process description

106 As illustrated in Fig. 1, besides the two membrane units MS1 and MS2, which are the `core' of 107 the studied separation process, the other main pieces of equipment are the compressors C1 and C2; 108 heat exchangers HEX1, HEX2, and HEX3; vacuum pumps VP1 and VP2; mixers M1 and M2; and 109 splitters SP1 and SP2. The incoming gas mixture feed F_0 increases its pressure in C1 and reaches the 110 operating temperature T_{MS1} in HEX1. Then, it is optionally mixed in M1 with a fraction of the 111 retentate stream obtained in MS1 (RRMS1) and/or a fraction of the retentate obtained in MS2 112 (RRMs2_MS1). Afterward, the resulting stream is fed to MS1 obtaining two streams: the permeate 113 stream (which is enriched in H_2) and the retentate stream. The permeate membrane side operates 114 under vacuum whether a vacuum pump VP1 is used to create the driving force for component 115 separation. The permeate stream PMS1 leaving the VP1 decreases its temperature in HEX2 and 116 increases its pressure in C2. Afterward, it reaches the operating temperature T_{MS2} in HEX3 and can 117 be optionally mixed in M2 with a fraction of the retentate obtained in MS2 (RRMs2). Finally, the 118 resulting stream is fed to MS2 obtaining the retentate and permeate streams corresponding to this 119 stage.

120



121 122

Figure 1. Schematic of the studied two-stage membrane process configuration.

123 The driving force for component permeation can be created in different ways: (i) by 124 compressing the feed F₀ by means of C1 in the first stage and compressing the permeate P_{MS1} by 125 means of C2 in the second stage (i.e. no vacuum is applied at the permeate side of the membranes); 126 (ii) by applying vacuum at the permeate sides (i.e. no compression of the feed and permeate streams 127 is applied); and (iii) by combining both compression and vacuum. But the best way to create the 128 driving force depends on other factors such as membrane areas, costs, and design specifications. The 129 higher compressor pressure ratio the smaller membrane area and the higher H₂ purity, but the 130 higher power requirement to run the compressor. Similarly, the higher pressure ratio of a vacuum 131 pump the smaller membrane area and the higher H₂ purity, but the higher vacuum level to run the 132 vacuum pump. In both cases, the optimal operating pressure values depend on the relationships 133 between investment and operating costs. Thus, it is clear the importance of optimizing 134 simultaneously all the techno-economic trade-offs that exist between the process variables.

135 **3. Problem statement**

136 The problem stated in this work is the optimal separation of H_2 from a given CO₂/CO/H₂/N₂ gas 137 mixture by means of the two-stage membrane process described above, to attain an H₂ recovery 138 target level of 90% varying parametrically the H_2 product purity between 0.90 and 0.95, by 139 minimization of the total annual cost, based on a NLP formulation. Both the permeability and 140 selectivity of the membrane material are assumed to be known and are taken from the literature [7]. 141 As a result, the minimum total annual cost, the optimal process-unit sizes (membrane unit 142 areas, heat exchanger areas, compressor and vacuum-pump power capacities), the optimal 143 operating conditions (pressure, temperature, flow rate, and composition of the retentate and 144 permeate streams), as well as the optimal values of the cost components (total investment, 145 individual process-unit acquisition cost, capital and operating expenditures, among others), are 146 provided.

147 4. Process modeling

148 4.1. Assumptions and process mathematical model

149 Briefly, some main assumptions considered for modeling the membrane unit are the following: 150 the permeability values correspond to the pure species and are not influenced by the operating 151 pressure; steady state behavior; plug-flow pattern; constant total pressure at each membrane side. 152 Regarding the mathematical modeling, the component mass balance in the membrane module is 153 described by a set of algebraic equations obtained by discretization of the resulting set of ordinary 154 differential equations employing the backward finite difference method (BFDM). The mathematical 155 model that describes the process flow sheet illustrated in Fig. 1 involves nonlinear constraints due to 156 the presence of bilinear terms in the mass and energy balances (multiplications of concentrations 157 and flow rates, and multiplications of enthalpies and flow rates) as well as equations to calculate 158 costs. A complete list of the assumptions and the mathematical model used in this study to describe 159 the membrane stages and the other process units (compressors, heat exchangers, etc.) can be found 160 in Arias [11] and Arias et al. [8] and are also provided in Appendix A. Next, the considered cost 161 model is presented.

162 4.2. Cost model

163 The total annual cost (TAC, in M\$ yr.⁻¹), capital expenditures (CAPEX, in M\$), annualized 164 capital expenditures (annCAPEX, in M\$ yr.⁻¹), and operating expenditures (OPEX, in M\$ yr.⁻¹) are 165 calculated by Eqs. (1–5).

$$TAC = annCAPEX + OPEX$$
(1)

$$annCAPEX = CRF \cdot CAPEX$$
(2)

$$CRF = \frac{i \cdot (1+i)^{n}}{(1+i)^{n} - 1}$$
(3)

$$CAPEX = f_1 \cdot C_{INV}$$
(4)

$$OPEX = f_2 \cdot C_{INV} + f_3 \cdot OLM + f_4 \cdot C_{RM}$$
(5)

166 In Eq. (5), OLM accounts for manpower and maintenance costs. A detailed calculation of the 167 economic factors f_1 (4.98), f_2 (0.464), f_3 (2.45), and f_4 (1.055) can be found in [8], which were estimated 168 based on the guidelines given in [12] and [13].

169 The total investment cost (C_{INV} , in M\$) is calculated by Eq. (6), where the investment costs of the 170 individual process units are estimated by Eqs. (7–10):

$$C_{INV} = I_{C1} + I_{C2} + I_{VP1} + I_{VP2} + I_{HEX1} + I_{HEX2} + I_{HEX3} + I_{MS1} + I_{MS2}$$
(6)

$$I_{HEX} = 0.3574 \cdot \left(\frac{A_{HEX}}{929}\right)^{0.6}$$
; HEX: HEX1, HEX2, HEX3 (7)

$$I_{\rm C} = 2.7878 \cdot \left(\frac{W_{\rm C}}{2000}\right)^{0.6}$$
; C: C1, C2 (8)

$$I_{VP} = 2.25034 \ 10^{-6} \cdot W_{VP}; VP: VP1, VP2$$
 (9)

$$I_{MS} = 5.28034 \ 10^{-5} \cdot A_{MS} + 0.24884 \cdot \left(\frac{0.1}{55} \cdot P^{H}\right)^{0.875} \cdot \left(\frac{A_{MS}}{2000}\right)^{0.7}; \text{ MS: MS1, MS2}$$
(10)

171 The raw material and utility cost (C_{RM} , in M\$ yr.⁻¹) used in Eq. (5) is calculated by Eq. (11). It 172 depends on the cost of electricity (C_{EP}), cooling water (C_{CW}), and membrane replacement (C_{MR}),

depends on the cost of electricity (CEP), cooling water (CCW), and membrane replacement (CMR),
 which are expressed by Eqs. (12–14), respectively:

$$C_{\rm RM} = C_{\rm EP} + C_{\rm CW} + C_{\rm MR} \tag{11}$$

$$C_{EP} = cru_{EP} \cdot TW \cdot OT \tag{12}$$

$$C_{CW} = cru_{CW} \cdot F_{CW} \cdot 1.8 \times 10^{-2} \cdot (3600 \cdot OT)$$
(13)

$$C_{MR} = 0.2 \cdot cru_{MR} \cdot (A_{MS1} + A_{MS2})$$
(14)

where the specific costs cru_{EP}, cru_{cw}, and cru_{MR} are, respectively, 0.072 \$ kW⁻¹, 0.050929 \$ kg⁻¹, and
 10.0 \$ m⁻². An operation period (OT) of 6570 h yr.⁻¹ was considered.

The complete model describing the entire process contains 3952 equations (equality and inequality constraints) and 3274 variables. It was implemented in GAMS –General Algebraic Modeling System– [14] and solved with CONOPT which is based on the generalized reduced gradient algorithm [15]. Therefore, global optimality of the discussed solutions cannot be guaranteed because of the use of a local search NLP solver. To guarantee it, a global optimization algorithm must be used instead.

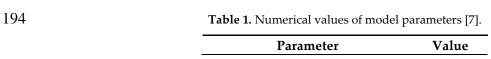
182 5. Results and discussion

The values of the model parameters used are taken from [7], which are listed in Table 1. In Subsection 5.1, optimal solutions obtained by minimizing the total annual cost for a variation range of H₂ product purity are discussed. In Subsection 5.2, optimal solutions obtained by minimizing the total membrane area and the electric power are presented and compared to the obtained by minimizing the total annual cost for the same design specifications.

188 5.1. Optimal solutions corresponding to the minimization of the total annual cost

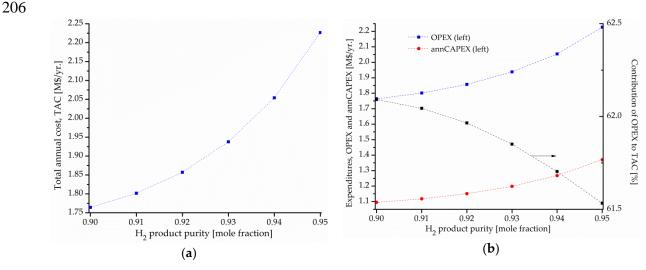
The main optimization results for a H₂ recovery target level of 90% and H₂ product purity target
 values in the range 0.90–0.95 are presented in Figs. 2–10.

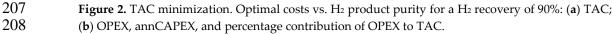
- 191
- 192
- 193



Feed specification				
Flow rate, mol s ⁻¹	0.02777			
Temperature, K	313.15			
Pressure, MPa	0.10132			
Composition (mole fraction)				
CO ₂	0.04			
CO	0.16			
H ₂	0.18			
N2	0.62			
Membrane material (Polymer)				
Permeance, mole m ⁻² s ⁻¹ MPa ⁻¹)				
CO ₂	8.4441 10-3			
СО	7.4571 10-4			
H2	2.8710 10-2			
N2	4.0781 10-4			

196 Figure 2a shows that the total annual cost (TAC) value increases as the H₂ purity increases since 197 the H₂ permeate flow rate in the first stage increases to satisfy the increased purity, as shown in Figs. 198 7–9 for H₂ product purity levels of 0.90, 0.91, and 0.94, respectively. For instance, compared to 0.90 199 H₂ product purity, the minimum TAC value obtained for 0.95 H₂ purity increases by 26.2%, -from 200 1.76421 to 2.22688 M\$ yr.⁻¹- as consequence of the increase of both the OPEX value by 25.1% -from 201 1.09542 to 1.37026 M\$ yr.⁻¹- and the annCAPEX value by 28.1% -from 0.66879 to 0.85663 M\$ yr.⁻¹-. 202 On the other hand, it can be observed in Fig. 2b that the contribution ratio between OPEX and 203 annCAPEX to the TAC remains almost constant with increasing H₂ purity values. Certainly, the 204 contribution of OPEX to the TAC varies slightly from 62.1% to 61.6% for H₂ product purity values of 205 0.90 and 0.95, respectively.

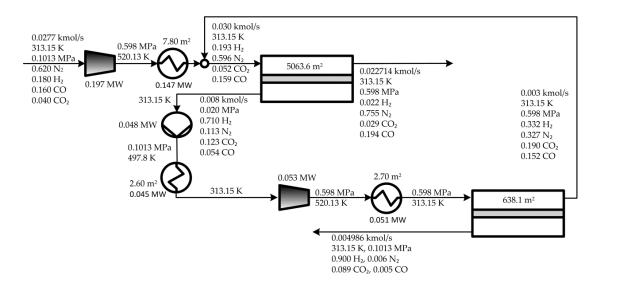


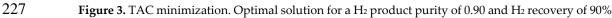


Also, Fig. 2a shows that an increment of H₂ product purity in 0.01 determines different percentage variations in costs depending on the purity level itself. For instance, an increment of H₂ purity from 0.90 to 0.91 implies an increase of TAC in 0.03739 M\$ yr.⁻¹ (from 1.76421 to 1.80160 M\$ yr.⁻¹, i.e. 2.1%) while the same increment from 0.94 to 0.95 implies an increase of TAC in 0.17274 M\$ yr.⁻¹ (from 2.05414 to 2.22688 M\$ yr.⁻¹, i.e. 8.4%).

By comparing Figs. 3 and 4 it can be observed that an increment of H_2 purity from 0.90 to 0.94 increases the permeate flow rate –from 0.008 to 0.013 kmol s⁻¹– and H_2 concentration –from 0.710 to

- 0.774 mole fraction- in the first stage, but decreases the permeate flow rate in the second stage -from 4.986x10⁻³ to 4.774x10⁻³ kmol s⁻¹-. It is interesting to note that, in order to reach these flow rate values and H₂ purities, the electric power requirement by compressors and vacuum pump increases in total 0.108 MW (0.026 MW, 0.049 MW, and 0.033 MW in C1, C2, and VP1, respectively) while the optimal total membrane area decreases 332.1 m² (58.0 m² and 274.1 m² in the first and second stage, respectively). Thus, the optimal cost-based trade-offs indicate that it is more beneficial to increase the total electrical power -to operate the process at higher operating pressure values P^H as shown in Fig. 5- rather than to increase the total membrane area.





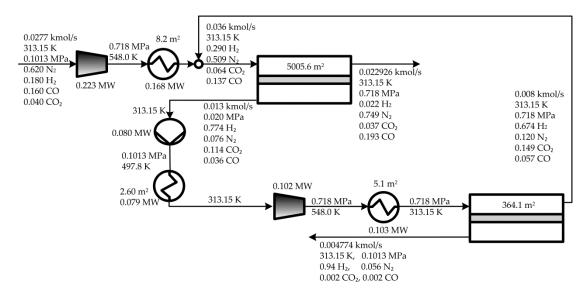


Figure 4. TAC minimization. Optimal solution for a H2 product purity of 0.94 and H2 recovery of 90%

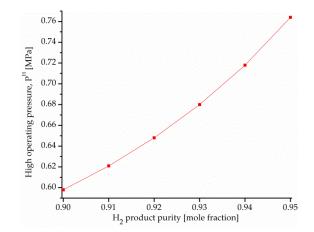
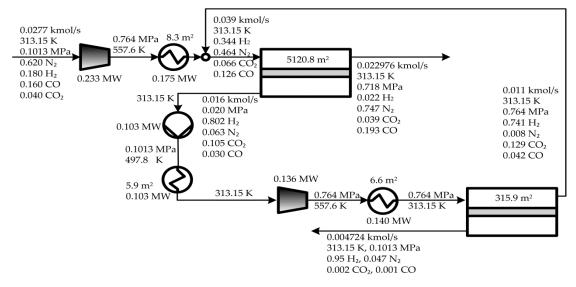




Figure 5. TAC minimization. Optimal high operating pressure P^H (retentate side) versus H₂ product
 purity.

241



242

243 Figure 6. TAC minimization. Optimal solution for a H₂ product purity of 0.95 and H₂ recovery of 90%

244 However, the comparison of Figs. 4 and 6 shows that a different trade-off is established 245 between the required total electric power and total membrane area when the H_2 product purity is 246 increased from 0.94 to 0.95. In this case, it is necessary to increase both the total membrane area 247 about 67.0 m² -from 5369.7 to 5436.7 m² - and the electric power about 0.065 MW -from 0.406 to 0.471 248 MW- in order to satisfy a desired H_2 purity of 0.95. It is interesting to note that the increase of the 249 total membrane area results from an increase of the area of the first stage in 115.2 m² and a decrease 250 of the area of the second stage in 48.2 m^2 , which is a trend opposite to the one observed when the H₂ 251 purity increases from 0.90 to 0.94 (Figs. 3 and 4), where the area of MS1 and MS2 decreases and 252 increases, respectively, with increasing purity levels. This behavior can be better understood by 253 observing in Fig 7 the individual variation of the area of both membranes with increasing product 254 purity levels. The figure interestingly shows that the curve of the membrane area corresponding to 255 MS1 has a minimum value at a H₂ purity value of 0.93, and that the one corresponding to MS2 256 decreases practically linearly in the studied purity variation range. This is one of the reasons of why 257 dissimilar trade-offs between the same process variables are established at different values of H₂ 258 purity levels.

- 259
- 260

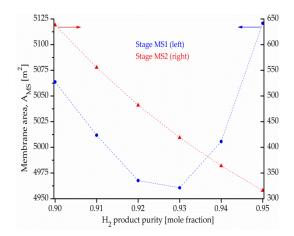




Figure 7. TAC minimization. Optimal membrane areas of stages MS1 and MS2 versus H₂ product
 purity.

These results clearly show the advantages of having an optimization mathematical model, mainly when both high recovery and purity levels are simultaneously targeted using membranes in gas separation processes. Certainly, it allows to identify the critical trade-offs that are otherwise difficult to distinguish at first glance.

269 Regarding the optimal contribution of each process unit to the total investment, Fig. 8a shows 270 that the compressor C1 used for compressing the feed in the first stage is the largest contributor; its 271 contribution increases practically linearly with increasing H₂ product purity values. It is followed by 272 the compressor C2 used in the second stage for compressing the permeate leaving the vacuum pump 273 VP1. In contrast to C1, the contribution of C2 increases exponentially with increasing H_2 product 274 purity values, showing a behavior similar to the one observed for the optimal high operating 275 pressure P^{H} values (Fig. 5). The third contributor to the total investment is the membrane area 276 required in the first stage AMS1, with an investment that remains almost constant with increasing H2 277 product purities. The vacuum pump VP1 is the fourth contributor, whose investment increases more 278 importantly at high H₂ purity values. The contributions of the remaining process units are 279 comparatively less important or practically insignificant.



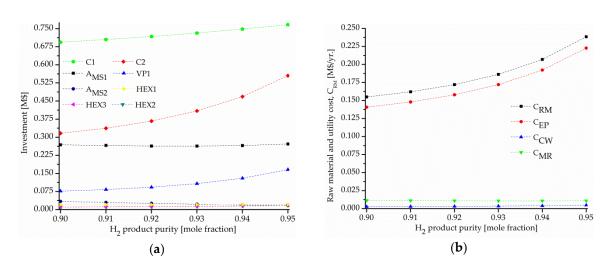


Figure 8. TAC minimization. Optimal costs vs. H2 product purity for a H2 recovery of 90%: (a)
 Process-unit investments; (b) Raw material and utility cost CRM, with cost for electric power EP,
 cooling water CW, and membrane replacement MR.

Regarding the distribution of raw material and utility cost CRM shown in Fig. 8b, the cost of electricity for running the compressors and the vacuum pump is by far the major contributor, and it increases more rapidly with increasing H₂ product purity levels.

289

Figure 9 clearly shows that the increases in electric power required by the compressor of the second stage C2 and the vacuum pump VP1 are more significant than the increase in electric power required by compressor C1 in the first stage.

293

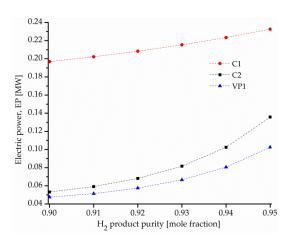
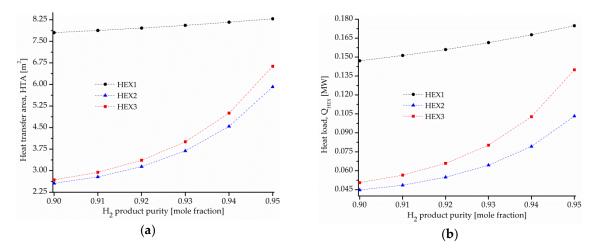


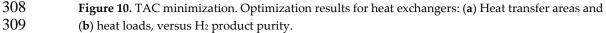


Figure 9. TAC minimization. Optimal sizes of the main process units versus H₂ product purity: (a)
 Compressors and vacuum pump; (b) Membrane areas of stages MS1 and MS2.

297

298 Finally, Fig. 10 shows the heat transfer areas of HEX2 and HEX3 (Fig. 10a) and their 299 corresponding heat loads (Fig. 10b) increase in greater proportion than the ones corresponding to 300 HEX1 with increasing H₂ purity levels. The area increases in HEX1 and HEX3 are mainly due to the 301 increase in the compression ratio of the compressors which rises their outlet temperatures, thus 302 requiring more area for heat transfer to reach the operating temperature of the stages (313.15 K). On 303 the other hand, the increase of the heat transfer area of HEX2 –located after the vacuum pump VP1– 304 and its corresponding heat load is only due to the increase in the permeate flow rate since the first 305 stage operates with a vacuum level of 0.020 MPa, what implies the same pressure ratio (5.065) and, 306 therefore, the same output temperature (497.8 K) for all the H₂ product purity levels. 307





310

311 5.2. Influence of the objective functions on the optimal design and operating conditions

312 As introduced earlier, the aim of this section is to study the influence of the objective functions 313 that are minimized on the optimal solutions and to see how these solutions can be properly used in 314 global optimization algorithms. To this end, the same process and cost mathematical models are 315 used to solve the optimization problems that result by considering the two remaining objective 316 functions: minimization of the total membrane area TMA (Eq. A32) and minimization of the total 317 power TW (Eq. A34). The optimization results obtained for the three objective functions are 318 compared in Table 2 -in terms of costs- and Table 3 -in terms of process-unit sizes and operating 319 conditions-. The optimal solutions resulting from the minimization of TAC, minimization of TMA,

320 and minimization of TW are hereafter referred as `osTAC', `osTMA', and `osTW', respectively.

Cost item	Min TMA	Min. TAC	Min TW
	osTMA	osTAC	osTW
TAC (M\$ yr-1)	1.85056	1.76421	2.11552
OPEX (M\$ yr1)	1.15843	1.09542	1.26216
annCAPEX (M\$ yr1)	0.69214	0.66879	0.85336
Cinv (M\$)	1.48076	1.43082	1.82568
Ic1	0.85171	0.69360	0.48869
Ic2	0.36483	0.31653	0.27448
Ima_ms1	0.13376	0.26859	0.83129
Ivp1	0.06926	0.07670	0.10428
IHEX1	0.02188	0.02031	0.01799
Ima_ms2	0.01843	0.03398	0.08489
Інехз	0.01101	0.01069	0.01144
IHEX2	0.00987	0.01041	0.01261
Скм (М\$ уг1)	0.19269	0.15497	0.13906
Ce	0.18332	0.14077	0.10236
Cmr	0.00571	0.01140	0.03463
Ccw	0.00366	0.00279	0.00206

321 Table 2. Minimization of TAC, TMA, and TW for 90% H₂ recovery and 0.90 H₂ product purity: Costs

Table 3. Minimization of TAC, TMA, and TW for 90% H2 recovery and 0.90 H2 product purity:Process-unit sizes and operating conditions.

	-	-	
Cost item	Min TMA	Min. TAC	Min TW
	osTMA	osTAC	osTW
TMA (m ²)	2854.23	5701.66	17316.96
MA _{MS1}	2510.80	5063.60	15714.90
MA _{MS2}	343.43	638.06	1602.06
TW (MW)	0.38754	0.29759	0.21639
W _{C1}	0.27717	0.19684	0.10981
Wc2	0.06746	0.05325	0.04199
W_{VP1}	0.04290	0.04751	0.06459
HTAHEX1 (m ²)	8.83941	7.80177	6.37764
HTAHEX2 (m ²)	2.305	2.6	3.5

322 323

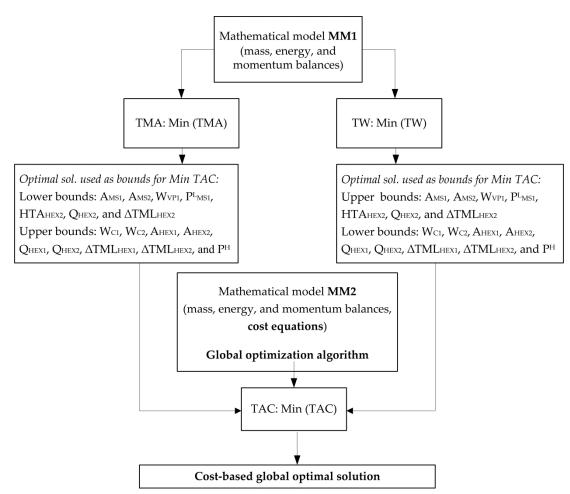
HTAHEX3 (m ²)	2.81339	2.68000	2.99720
QHEX1 (MW)	0.20996	0.14713	0.08090
$Q_{\text{HEX2}}(MW)$	0.041	0.045	0.060
QHEX3 (MW)	0.06682	0.05054	0.03802
ΔTML Hex1 (K)	85.533	67.908	45.676
ΔTML Hex2 (K)	62.871	62.871	61.236
ΔTML HEX3 (K)	85.533	67.908	45.676
P ^H (MPa)	1.01320	0.59834	0.30396
PL _{MS1} (MPa)	0.02000	0.02000	0.02102
PL _{MS2} (MPa)	0.10132	0.10132	0.10132

325 Table 2 shows that the TAC value obtained in osTAC is 4.7% and 16.6% lower than those 326 obtained in osTMA and osTW, respectively; while the OPEX value obtained in osTAC is 5.4% and 327 13.2% lower than the respective ones obtained in osTMA and osTW. The annCAPEX value obtained 328 in osTAC is 3.4% and 21.6% lower than the obtained in osTMA and osTW, respectively. Table 3 329 shows that the minimal TMA value required by the process in solution osTMA is 2854.23 m², which 330 is, respectively, 49.9% and 83.5% lower than the value obtained in osTAC and osTW, but the electric 331 power required in osTMA is 30.2% and 79.1% higher than the required in osTAC and osTW, 332 respectively. Table 3 shows that the minimization of TMA (2854.23 m²) implies the highest TW value 333 (0.38754 MW) reaching the upper bound for P^H (1.01320 MPa). Also, Table 3 shows that the 334 minimization of TW (0.21639 MW) implies the highest TMA value (17316.96 m²).

335 So far, the optimal solutions obtained by minimization of three different objective functions by 336 using a local search algorithm have been presented. Next, the features of these solutions are 337 considered to propose a systematic strategy for providing bounds for model variables in global 338 optimization (GO) methods. The application of GO algorithms using deterministic mathematical 339 models allows to obtain a solution for a given global optimality tolerance. As is well known from a 340 computational point of view, the calculation of good lower and upper bounds is crucial for the 341 success of any GO algorithm [16]. Sherali et al. [17] proposed an improved method to develop tight 342 linear relaxations to calculate global lower bounds for a design problem associated with the water 343 distribution network. To this end, they exploited the characteristics (nature) of the nonlinear 344 constraints such as monotonicity and the convex-concave functions. The proposed method allowed 345 to obtain novel network designs as well as to identify promising search sub-regions. Ruiz and 346 Grossmann [18] proposed an efficient procedure to find strong bounds in generalized disjunctive 347 (GDP) problems. Based on the theory associated with disjunctive programming, they proposed 348 several rules to generate more efficiently new relaxations to predict strong bounds for the global 349 optimum. The proposed procedure considerably enhanced the computational cost because it leads 350 to a significant reduction of the number of nodes to evaluate by the branch-and-bound methods. 351 Kirst et al. [19] presented a comprehensive discussion about the difficulties to determine good 352 bounds in branch-and-bound methods for GO of mathematical models involving non-convex 353 constraints. Also, they proposed a consistent way to calculate bounds by perturbing infeasible points 354 to feasible ones by iterating along Mangasarian-Fromovitz directions. The perturbations were done 355 using optimal solutions obtained from linear optimization problems.

356 Taking into consideration the above comments and the characteristics of the optimal solutions 357 reported in Tables 2 and 3, it is interesting to investigate how the information provided by the 358 optimal solutions corresponding to the minimization of the total membrane area (osTMA) and the 359 total power (osTW) can be used in GO algorithms to solve the problem of minimization of the total 360 annual cost, and therefore, to guarantee the optimality of the optimal solutions for the two-stage 361 membrane process for H₂ separation studied in this paper –which in fact can be straightforward 362 applied to other processes-. More precisely, in order to determine tight variable bounds, the idea 363 behind is to use the information predicted by the same model of the process but considering two

different situations: membrane area minimization and power minimization, which represent two extremes that can be used efficiently to narrow the feasible region of the cost optimization problem. Unlike other works, it is here intended to establish a systematic bounding procedure using information inherent to the process obtained in a rational way instead of exploiting the nature of the associated constraints –at the beginning of the methodology– without applying any rational criterion.



371

Figure 11. Schematic of a variable bounding and solution strategy proposed for solving the costoptimization problem via global optimization.

374 By comparing numerically the results presented in Tables 2 and 3, it can be observed that the 375 optimal values corresponding to osTMA and osTW provide `good' lower/upper bounds for decision 376 variables to obtain the solution osTAC to global optimality. Thus, it is possible to propose a 377 pre-processing' phase to systematically and rationally provide good lower and upper bounds to 378 solve the cost minimization problem to global optimality. The schematic of the proposed 379 pre-processing methodology (Fig. 11) shows that two optimization problems -which only differ on 380 the objective function- are solved by using a same mathematical model of the process without 381 including cost constraints (model MM1). The first step is to solve the minimization of TMA; the 382 optimal values of Ams1, Ams2, WVP1, PLMS1, HTAHEX2, QHEX2, and ΔTMLHEX2 are provided as lower 383 bounds to solve the minimization of TAC while the optimal values of WC1, WC2, AHEX1, AHEX2, QHEX1, 384 QHEX2, ΔTML HEX1, ΔTML HEX2, and P^H are provided as upper bounds. The second step consists on the 385 minimization of TW; the optimal values of WC1, WC2, AHEX1, AHEX2, QHEX1, QHEX2, ATMLHEX1, ATMLHEX2, 386 and P^H are now provided as lower bounds while the values of AMS1, AMS2, WVP1, P^LMS1, HTAHEX2, QHEX2, 387 and ΔTML_{HEX2} as upper bounds. Thus, if the minimization of TMA provides a lower bound for a 388 given decision variable, then the minimization of TW provides an upper bound for it, and vice versa. 389 It is said that the pre-processing phase provides bounds in a rational way because they represent

390 limits for the sizes of the pieces of equipment and/or process operation conditions. They can be used 391 to identify smaller search spaces for the cost optimization problem and reduce the number of 392 iterations, and consequently, the computing time. In addition, one of the two solutions (osTMA or 393 osTW) can be used as an initial guess point in the global optimization algorithm because they are

394 both feasible solutions for the cost optimization problem, thus facilitating the model convergence.

395 6. Conclusions

This paper presented the optimization results of a two-stage membrane system for H₂ separation by minimization of the total annual cost, the total membrane area, and the total electric power as single objective functions, employing a nonlinear mathematical model implemented in GAMS environment.

400 First, a detailed discussion of optimal solutions obtained by minimizing the total annual cost for 401 desired H₂ product purity values ranging between 0.90 and 0.95 and keeping constant the H₂ 402 recovery at 90% was presented. One of the optimization results showed that an increment of H₂ 403 product purity in 0.01 determines different percentage variations in costs depending on the purity 404 level itself. For instance, an increment of H₂ purity from 0.90 to 0.91 implies an increase of the total 405 annual cost in 0.03739 M\$ yr.-1 (2.1%) while the same increment from 0.94 to 0.95 implies an increase 406 in 0.17274 M\$ yr.-1 (8.4%). Also, the optimal trade-off existing between the total membrane area and 407 the total electric power depends on the values of H₂ purity. In fact, it was found that different 408 trade-offs are established between the required total electric power and the total membrane area 409 when the H₂ product purity is increased from 0.90 to 0.94 and from 0.94 to 0.95. In the former case, 410 the total electric power increases and the total membrane area decreases with the increasing of the 411 H₂ purity. The optimal cost-based trade-offs indicated that it is more beneficial to increase the total 412 electric power -to operate the process at higher operating pressure values- rather than to increase 413 the total membrane area. However, in the last case, it was observed that both the total membrane 414 area and the total electric power increase with the increasing of the H₂ product purity.

415 The proposed mathematical model was solved considering the three aforementioned objective 416 functions, and the obtained solutions were compared. From this comparison, it was observed that 417 the optimal values obtained by minimizing the total membrane area and the total electric power -as 418 single objectives- are 'good' bounds when the total annual cost is intended to be minimized to 419 global optimality. By inspecting the numerical value of each model variable and the objective 420 function, it is possible to propose a 'pre-processing' phase to systematically and rationally provide 421 good lower and upper bounds to solve the minimization of the total annual cost to global optimality. 422 In this work, a local search optimization algorithm was used for cost minimization. In a next paper, a 423 global optimization algorithm will be used instead, exploiting these solution features.

424 Author Contributions: All authors contributed to the analysis of the results and to writing the manuscript. 425 A.M.A and P.L.M developed and implemented in GAMS the generic mathematical model of a membrane 426 module and the cost model, and collected and analyzed data. S.F.M. developed and implemented in GAMS the 427 model of the whole process, and wrote the first draft of the manuscript. N.J.S improved the discussion of 428 results. J.A.C contributed with numerical and solution issues. M.C.M. conceived and supervised the research 429 and provided feedback to the content.

Acknowledgments: The financial support from the Consejo Nacional de Investigaciones Científicas y Técnicas
 (CONICET), the Agencia Nacional de Promoción Científica y Tecnológica (ANPCyT), and the Facultad
 Regional Rosario of the Universidad Tecnológica Nacional (UTN) from Argentina is gratefully acknowledged.

433 **Conflicts of Interest:** The authors declare no conflict of interest.

434 Appendix A: Process mathematical model

435 A.1. Main model assumptions

- 436 The assumptions considered for modeling the membrane units are [8]:
- All components can permeate.

- The component permeability is not affected by the operating pressure.
- The pressure drop is negligible at both membrane sides.
- The pressure of the feed and retentate streams is the same.
- Plug-flow pattern is considered at both membrane sides.
- Each membrane module operates isothermally.
- The Fick's first law is used.

444 A.2. Mathematical model

445 A.2.1. Mass balances

453

Figure A1a and A1b schematize the process configuration and the membrane module,
respectively, that are modeled. The equations describing the mass balance of component i in the
membrane module MS1, by applying the backward finite difference discretization method, are:

$$\frac{1}{2} \frac{(J-1)}{A_{MS1}} \left(-F_{MS1,j} \cdot x_{MS1,i,j} + F_{MS1,j+2} \cdot x_{MS1,i,j+2} \right) + \xi_i \left(P^H \cdot x_{MS1,i,j+1} - P^L_{MS1} \cdot y_{MS1,i,j+1} \right) = 0; \quad j = 0, \quad \forall i \quad (A1)$$

$$\frac{1}{2} \frac{(J-1)}{A_{MS1}} \cdot \left(F_{MS1,j} \cdot x_{MS1,i,j} - 4 \cdot F_{MS1,j+1} \cdot x_{MS1,i,j+1} + 3 \cdot F_{MS1,j+2} \cdot x_{MS1,i,j+2} \right) +$$

$$\xi_{i} \cdot \left(P^{H} \cdot x_{MS1,i,j+2} - P_{MS1}^{L} \cdot y_{MS1,i,j+2} \right) = 0; \ j = 0, \dots, J-2$$

$$(A2)$$

449 ξ_i and A_{MS1} are the permeance of component i and membrane surface area, respectively. P^H and P^L_{MS1}

450 are the operating pressures in the retentate and permeate sides, respectively. The index j refers to a

451 discretization point which varies from 0 to 19 (J=19, i.e. 20 discretization points is considered).

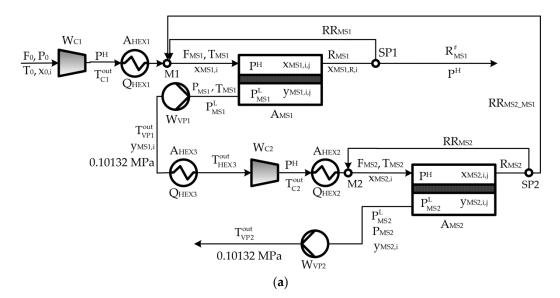
$$F_{MS1,j=0} \cdot x_{MS1,i,j=0} - F_{MS1,j=J} \cdot x_{MS1,i,j=J} - P_{MS1,j=0} \cdot y_{MS1,i,j=0} = 0; \ \forall i$$
(A3)

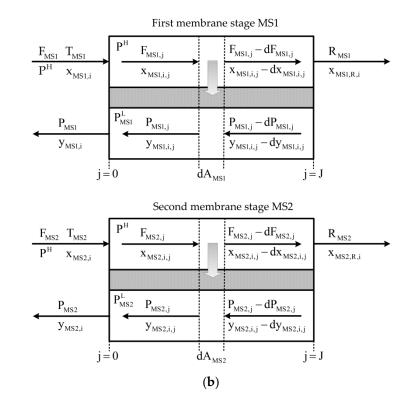
$$F_{MS1,j=0} - F_{MS1,j=J} - P_{MS1,j=0} = 0$$
(A4)

$$\sum_{i} x_{MS1,i,j} = 1; \quad \forall j, \ i = H_2, N_2, CO_2, CO$$
(A5)

$$\sum_{i} y_{MS1,i,j} = 1; \quad \forall j, \ i = H_2, N_2, CO_2, CO$$
(A6)

452 The second membrane stage MS2 is modeled in a similar way.





456 Figure A1. Representation and nomenclature: (a) Whole process; (b) Membrane modules.

457 The mass balances in splitters SP1 and SP2 are:

$$\mathbf{R}_{\mathrm{MS1}} = \mathbf{R}\mathbf{R}_{\mathrm{MS1}} + \mathbf{R}_{\mathrm{MS1}}^{\sharp} \tag{A7}$$

$$R_{MS2} = RR_{MS2} + RR_{MS2_MS1}$$
(A8)

458 A.2.2. Power requirement

459 The electric power required by compressors (C1, C2) and vacuum pumps (VP1, VP2) are 460 calculated as follows:

$$W_{C1} = \frac{F_0}{\eta_C} \cdot \left(\frac{\gamma}{\gamma - 1}\right) \cdot R \cdot T_0 \cdot \left[\left(\frac{P^H}{P_0}\right)^{\frac{\gamma - 1}{\gamma}} - 1\right]$$
(A9)

$$\frac{T_{C1}^{out}}{T_0} = \left(\frac{P^H}{P_0}\right)^{\frac{\gamma-1}{\gamma}}$$
(A10)

$$W_{C2} = \frac{P_{MS1}}{\eta_{C}} \cdot \left(\frac{\gamma}{\gamma - 1}\right) \cdot R \cdot T_{HEX3}^{out} \cdot \left[\left(\frac{P^{H}}{P_{0}}\right)^{\frac{\gamma - 1}{\gamma}} - 1\right]$$
(A11)

$$\frac{T_{C2}^{out}}{T_{HEX3}^{out}} = \left(\frac{P^{H}}{P_{0}}\right)^{\frac{\gamma-1}{\gamma}}$$
(A12)

461 γ , η_c , and P₀ are the adiabatic expansion coefficient (1.4), efficiency (0.85), and atmospheric pressure 462 (0.1013 MPa), respectively.

$$W_{VP1} = \frac{P_{MS1}}{\eta_{C}} \cdot \left(\frac{\gamma}{\gamma - 1}\right) \cdot R \cdot T_{MS1} \cdot \left[\left(\frac{P_{0}}{P_{MS1}^{L}}\right)^{\frac{\gamma - 1}{\gamma}} - 1\right]$$
(A13)

$$\frac{T_{VP1}^{out}}{T_{MS1}} = \left(\frac{P_0}{P_{MS1}^L}\right)^{\frac{\gamma-1}{\gamma}}$$
(A14)

$$W_{VP2} = \frac{P_{MS2}}{\eta_{C}} \cdot \left(\frac{\gamma}{\gamma - 1}\right) \cdot R \cdot T_{MS2} \cdot \left[\left(\frac{P_{0}}{P_{MS2}^{L}}\right)^{\frac{\gamma - 1}{\gamma}} - 1\right]$$
(A15)

$$\frac{T_{\text{VP2}}^{\text{out}}}{T_{\text{MS2}}} = \left(\frac{P_0}{P_{\text{MS2}}^{\text{L}}}\right)^{\frac{\gamma-1}{\gamma}}$$
(A16)

463 A.2.3. Energy balances and transfer areas of heat exchangers

464 The energy balances in the heat exchangers and their heat transfer areas are calculated by Eqs.465 (A17-A25):

$$\mathbf{F}_{0} \cdot \mathbf{c}_{\mathrm{P}}^{\mathrm{g}} \cdot \left(\mathbf{T}_{\mathrm{C1}}^{\mathrm{out}} - \mathbf{T}_{\mathrm{MS1}}\right) = \mathbf{n}_{\mathrm{HEX1,cw}} \cdot \mathbf{c}_{\mathrm{P}}^{\mathrm{cw}} \cdot \left(\mathbf{T}_{\mathrm{HEX1,cw}}^{\mathrm{out}} - \mathbf{T}_{\mathrm{HEX1,cw}}^{\mathrm{inl}}\right)$$
(A17)

$$P_{MS1} \cdot c_P^g \cdot \left(T_{C2}^{out} - T_{MS2}\right) = n_{HEX2,cw} \cdot c_P^{cw} \cdot \left(T_{HEX2,cw}^{out} - T_{HEX2,cw}^{inl}\right)$$
(A18)

$$P_{MS1} \cdot c_P^g \cdot \left(T_{VP1}^{out} - T_{HEX3}^{out}\right) = n_{HEX3,cw} \cdot c_P^{cw} \cdot \left(T_{HEX3,cw}^{out} - T_{HEX3,cw}^{inl}\right)$$
(A19)

$$A_{HEX1} = \frac{F_{0} \cdot c_{P}^{g} \cdot (T_{C1}^{out} - T_{MS1})}{U \cdot LMTD_{HEX1}}$$
(A20)

$$A_{HEX2} = \frac{P_{MS1} \cdot c_P^g \cdot (T_{C2}^{out} - T_{MS2})}{U \cdot LMTD_{HEX2}}$$
(A21)

$$A_{\text{HEX3}} = \frac{P_{\text{MS1}} \cdot c_{\text{P}}^{\text{g}} \cdot \left(T_{\text{VP1}}^{\text{out}} - T_{\text{HEX3}}^{\text{out}}\right)}{U \cdot \text{LMTD}_{\text{HEX3}}}$$
(A22)

$$LMTD_{HEX1} = \frac{\left(T_{C1}^{out} - T_{HEX1,cw}^{out}\right) - \left(T_{MS1} - T_{HEX1,cw}^{inl}\right)}{\ln \frac{\left(T_{C1}^{out} - T_{HEX1,cw}^{out}\right)}{\left(T_{MS1} - T_{HEX1,cw}^{inl}\right)}}$$
(A23)

$$LMTD_{HEX2} = \frac{\left(T_{C2}^{out} - T_{HEX2,cw}^{out}\right) - \left(T_{MS2} - T_{HEX2,cw}^{inl}\right)}{\ln\frac{\left(T_{C2}^{out} - T_{HEX2,cw}^{out}\right)}{\left(T_{MS2} - T_{HEX2,cw}^{inl}\right)}}$$
(A24)

$$LMTD_{HEX3} = \frac{\left(T_{VP1}^{out} - T_{HEX3,cw}^{out}\right) - \left(T_{HEX3}^{out} - T_{HEX3,cw}^{inl}\right)}{\ln \frac{\left(T_{VP1}^{out} - T_{HEX3,cw}^{out}\right)}{\left(T_{HEX3}^{out} - T_{HEX3,cw}^{inl}\right)}}$$
(A25)

The parameter U is the overall heat transfer coefficient, which is assumed to be 277.7 10⁻⁴ MW dam⁻² K⁻¹ for all heat exchangers.

468 A.2.4. Connecting constraints

The constraints used to relate model variables defined inside and outside of the membranemodule MS1 are:

$$F_{MS1} = F_{MS1,j=0}$$
 (A26)

$$x_{MS1,i} = x_{MS1,i,j=0}$$
 (A27)

$$\mathbf{F}_{\mathrm{MS1, j=J}} = \mathbf{R}_{\mathrm{MS1}} \tag{A28}$$

$$\mathbf{x}_{\mathrm{MS1,i,j=J}} = \mathbf{x}_{\mathrm{MS1,R,i}} \tag{A29}$$

$$P_{MS1} = P_{MS1,j=0}$$
 (A30)

$$y_{MS1,i} = y_{MS1,i,j=0}$$
 (A31)

471 Similar constraints are necessary for the membrane stage MS2.

472 A.2.5. Performance variables

The total membrane area TMA, total heat transfer area THTA, and total power are calculated by Eqs. (A32–A34), respectively:

$$TMA = A_{MS1} + A_{MS2}$$
(A32)

$$THTA = A_{HEX1} + A_{HEX2} + A_{HEX3} + A_{HEX4}$$
(A33)

$$TW = W_{C1} + W_{C2} + W_{VP1} + W_{VP2}$$
(A34)

475 Nomenclature

- 476 AMS#: membrane area required in the membrane stage MS#, m².
- 477 annCAPEX: annualized capital expenditures, M\$ yr.⁻¹
- 478 CAPEX: capital expenditures, M\$.
- 479 CRF: capital recovery factor, yr.⁻¹
- 480 CRM: raw material and utility cost, M\$ yr.-1
- 481 crucw: specific cost of the cooling water, M\$ kg-1.
- 482 cruEE: specific cost of the electricity, M\$ kW-1.
- 483 crume: specific cost of the membrane replacement, M\$ m⁻².
- 484 Fo: feed flow rate, kmol s⁻¹.
- 485 F_{MS#}: feed flow rate in the membrane stage MS#, kmol s⁻¹.
- 486 IMS#: investment for membrane area of the stage MS#, M\$.

- 487 IHEX#: investment for the heat exchanger HEX#, M\$.
- 488 IVP#: investment for the vacuum pump VP#, M\$.
- 489 Ic#: investment for the compressor C#, M\$.
- 490 OPEX: operating expenditures, M\$ yr.⁻¹
- 491 P^H: high operating pressure (retentate side), MPa.
- 492 PMS#: permeate flow rate obtained in the membrane stage MS#, kmol s⁻¹.
- 493 PL_{MS#}: operating pressure in the permeate side of the membrane stage MS#, MPa.
- 494 R_{MS#}: retentate flow rate obtained in the membrane stage MS#, kmol s⁻¹.
- 495 TAC: total annual cost, M\$ yr.-1
- 496 To: feed temperature, K.
- 497 T^{out} C#: outlet temperature from the compressor C# associated with the membrane stage MS#, K.
- 498 T_{MS#}: operating temperature in the membrane stage MS#, K.
- 499 T^{out} HEX#: outlet temperature from the heat exchanger HEX#, K.
- 500 TW: total power, MW.
- 501 Wcr: power required by the compressor C# associated with the membrane stage MS#, MW.
- 502 WVP#: power required by the vacuum pump VP# in the membrane stage MS#, MW.
- $x_{i,0}$: mole fraction of component i in the feed stream, dimensionless.
- 504 x_{MS#,i}: inlet composition of the component i in the membrane stage MS#, dimensionless.
- 505 xms#,i,j: mole fraction of the component i in the retentate stream of the membrane stage MS# at the
- 506 discretization point j, dimensionless.
- 507 x_{MS#,R,i}: mole fraction of the component i in the retentate stream leaving the membrane stage MS#, 508 dimensionless.
- 509 $y_{MS\#,i}$: mole fraction of the component i in the permeate stream leaving the membrane stage MS#,
- 510 dimensionless.
- 511 y_{MS1,i,j}: mole fraction of the component i in the permeate stream of the membrane stage MS# at the
- 512 discretization point j, dimensionless.
- 513
- 514

515 References

- 516 1. Ahmad, F.; Lau, K.K.; Shariff, A.M.; Murshid, G. Process simulation and optimal design of membrane
 517 separation system for CO₂ capture from natural gas. *Comp. Chem. Eng.* 2012, 36, 119-128,
 518 DOI:10.1016/j.compchemeng.2011.08.002.
- 519 2. Chowdhury, M.H.M.; Feng, X.; Douglas, P.; Croiset, E. A new numerical approach for a detailed
 520 multicomponent gas separation membrane model and AspenPlus Simulation. *Chem. Eng. Technol.* 2005, 28
 521 (7), 773-782, DOI:10.1002/ceat.200500077.
- 522 3. Xu, J.; Wang, Z.; Zhang, C.; Zhao,S.; Qiao, Z.; Li, P.; Wang, J.; Wang, S. Parametric analysis and potential
 523 prediction of membrane processes for hydrogen production and pre-combustion CO₂ capture. *Chem. Eng.*524 *Sci.* 2015, *135*, 202-216, DOI:10.1016/j.ces.2015.04.033.
- 4. Giordano, L.; Roizard, D.; Bounaceur, R.; Favre, E. Energy penalty of a single stage gas permeation process
 for CO₂ capture in post-combustion: A rigorous parametric analysis of temperature, humidity and
 membrane performances. *Energy Procedia*, 2017, 114, 636-641, DOI:10.1016/j.egypro.2017.03.1206.
- 5. Turi, D.M.; Ho, M.; Ferrari, M. C.; Chiesa, P.; Romano, M.C. CO₂ capture from natural gas combined cycles 529 by CO₂ selective membranes. *Int. J. Greenh. Gas Con.* **2017**, *61*, 168-183, DOI:10.1016/j.ijggc.2017.03.022.
- 530 6. Ghasemzadeh, K.; Jafari, M.; Sari, A.H.; Babalou, A.A. Performance investigation of membrane process in
- 531 natural gas sweeting by membrane process: Modeling study. *Chem. Prod. Process Model.* **2016**, *11*(1), 23–27.

- 532 7. Zarca, L.; Urtiaga, A.; Biegler, L.T.; Ortiz I. An optimization model for assessment of membrane-based
 533 post-combustion gas upcycling into hydrogen or syngas. J. Memb. Sci. 2018, 563, 83-92,
 534 DOI:10.1016/j.memsci.2018.05.038.
- Arias, A.M.; Mussati, M.C.; Mores, P.L.; Scenna, N.J., Caballero, J.A.; Mussati, S.F. Optimization of multi-stage membrane systems for CO₂ capture from flue gas. *Int. J. Greenh. Gas Control* 2016, *53*, 371-390, DOI:10.1016/j.ijggc.2016.08.005.
- 8. Ramírez-Santos, A.A.; Bozorg, M.; Addis, B.; Piccialli, V.; Castel, C.; Favre, E. Optimization of multistage
 membrane gas separation processes. Example of application to CO₂ capture from blast furnace gas. J.
 Membr.Sci. 2018, 566, 346-366, DOI:10.1016/j.memsci.2018.08.024.
- 541 10. Ohs, B.; Lohaus, J.; Wessling, M. Optimization of membrane based nitrogen removal from natural gas. J.
 542 *Memb. Sci.* 2016, 498, 291-301, DOI:10.1016/j.memsci.2015.10.007.
- 543 11. Arias, A.M. Minimization of greenhouse gases (GHGs) emissions in the energy sector employing
 544 non-conventional technologies. PhD Thesis, Universidad Tecnológica Nacional, Argentina, 2017. (In
 545 Spanish).
- Abu-Zahra, M.R.M.; Niederer, J.P.M.; Feron, P.H.M.; Versteeg, G.F. CO₂ capture from power plants: Part
 II. A parametric study of the economical performance based on mono-ethanolamine, *Int. J. Greenh. Gas Control* 2007, 1, 135-142, DOI:10.1016/S1750-5836(07)00032-1.
- 13. Rao, A.B.; Rubin, E.S. A technical, economic, and environmental assessment of amine-based CO₂ capture
 technology for power plant greenhouse gas control. *Environ. Sci. Technol.* 2002, 36, 4467-4475, DOI:10.1021/es0158861.
- 552 14. GAMS Development Corporation. General Algebraic Modeling System *GAMS*, Release 24.2.1. 2013,
 553 Washington DC, USA. http://www.gams.com
- 15. Drud, A.S. CONOPT 3 solver manual **2012**, ARKI Consulting and Development A/S, Bagsvaerd, Denmark.
- Furanik, Y.; Sahinidis, N.V. Domain reduction techniques for global NLP and MINLP optimization.
 Constraints 2017, 22(3), 338–376, DOI: 10.1007/s10601-016-9267-5.
- 557 17. Sherali, H.D.; Totlani, E.; Loganathan, G.V. Enhanced lower bounds for the global optimization of water
 558 distribution. *Water Resour. Res.* 1998, 34 (7), 1831-1841, DOI:10.1029/98WR00907.
- Ruiz, J.P.; Grossmann, I.E. A new theoretical result for convex nonlinear generalized disjunctive programs
 and its applications. *Proceedings of the 22nd European Symposium on Computer Aided Process Engineering* 2012,
 1–20, 1197-1201.
- 562 19. Kirst, P.; Stein, O.; Steuerman, P. Deterministic upper bounds for spatial branch-and-bound methods in
 563 global minimization with nonconvex constraints. *TOP* 2015, 23(2), 591–616,
 564 DOI:10.1007/s11750-015-0387-7.
- 565